



FUNDACIÓN PARA EL
DESARROLLO DE LAS NUEVAS
TECNOLOGÍAS DEL HIDRÓGENO
EN ARAGÓN



国立研究開発法人
科学技術振興機構
Japan Science and Technology Agency



Producción de syngas en un módulo electroquímico para su posterior utilización con H₂ renovable en la producción de combustibles sintéticos

Syngas production by electrochemical reactor as a key step in the production of alternative fuels

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4AirCRAFT-Air Carbon Recycling for Aviation Fuel Technology



- **Funding Programme:** H2020-EU.3.3.3. – Alternative fuels and mobile energy sources
- **Topic:** LC-SC3-RES25-2020 – International cooperation with Japan for Research and Innovation on advanced biofuels and alternative renewable fuels



This project has received funding from the European Union's Horizon 2020 research and innovation programme under grant agreement No 101022633.



This work is supported by Japan Science and Technology Agency (JST) under Grant Agreement No JPMJSC2102.

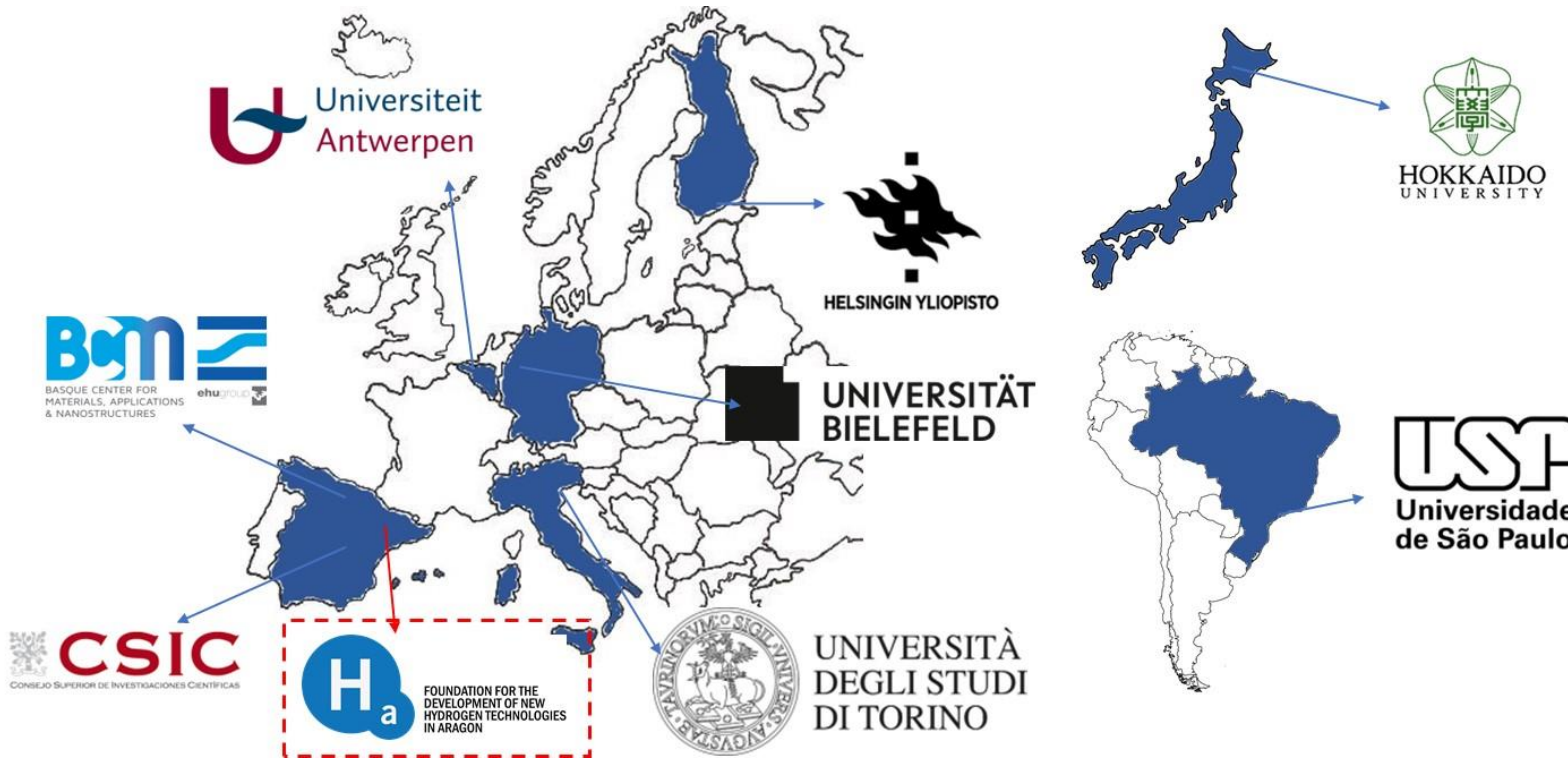


This project is developed in the frame of a Mission Innovation Challenge supported by The Sao Paulo Research Foundation (FAPESP) under Grant No. 2022/04751-0

4AirCRAFT Consortium



Coordination: Aragon Hydrogen Foundation (FHa, Spain)



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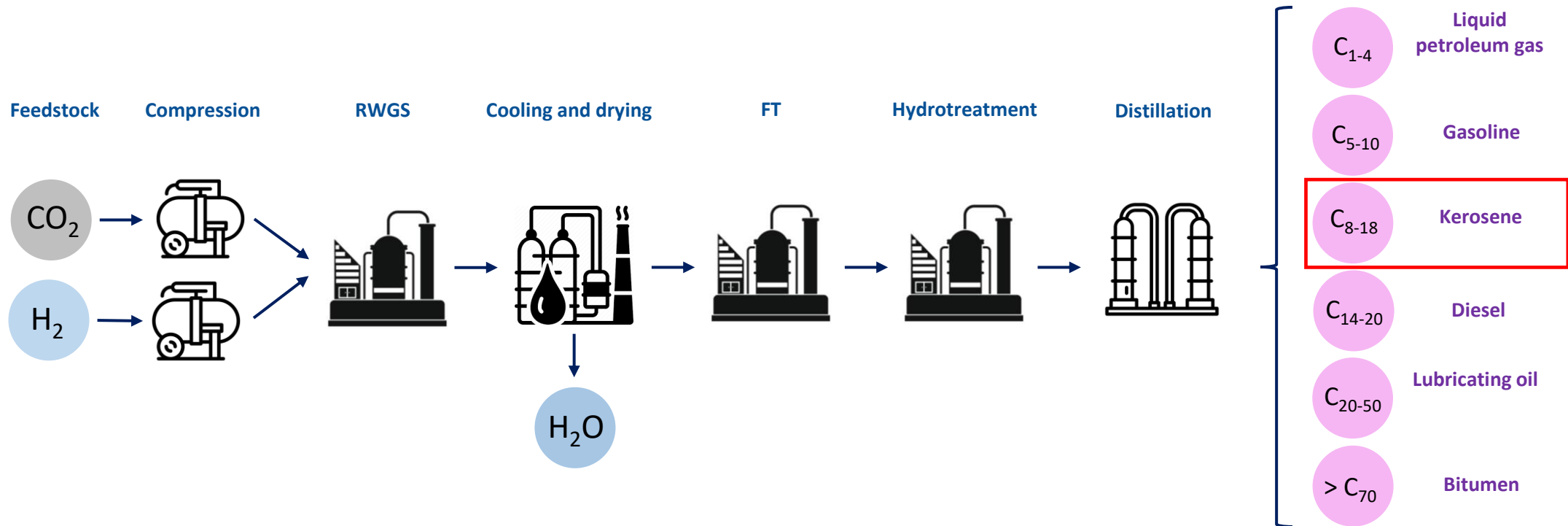
Advisory Panel group

- Sustainable fuels producers
- Petro-chemical industry
- Chemical industry
- CO₂ producers
- Logistics, Ports, etc.



What is our motivation?

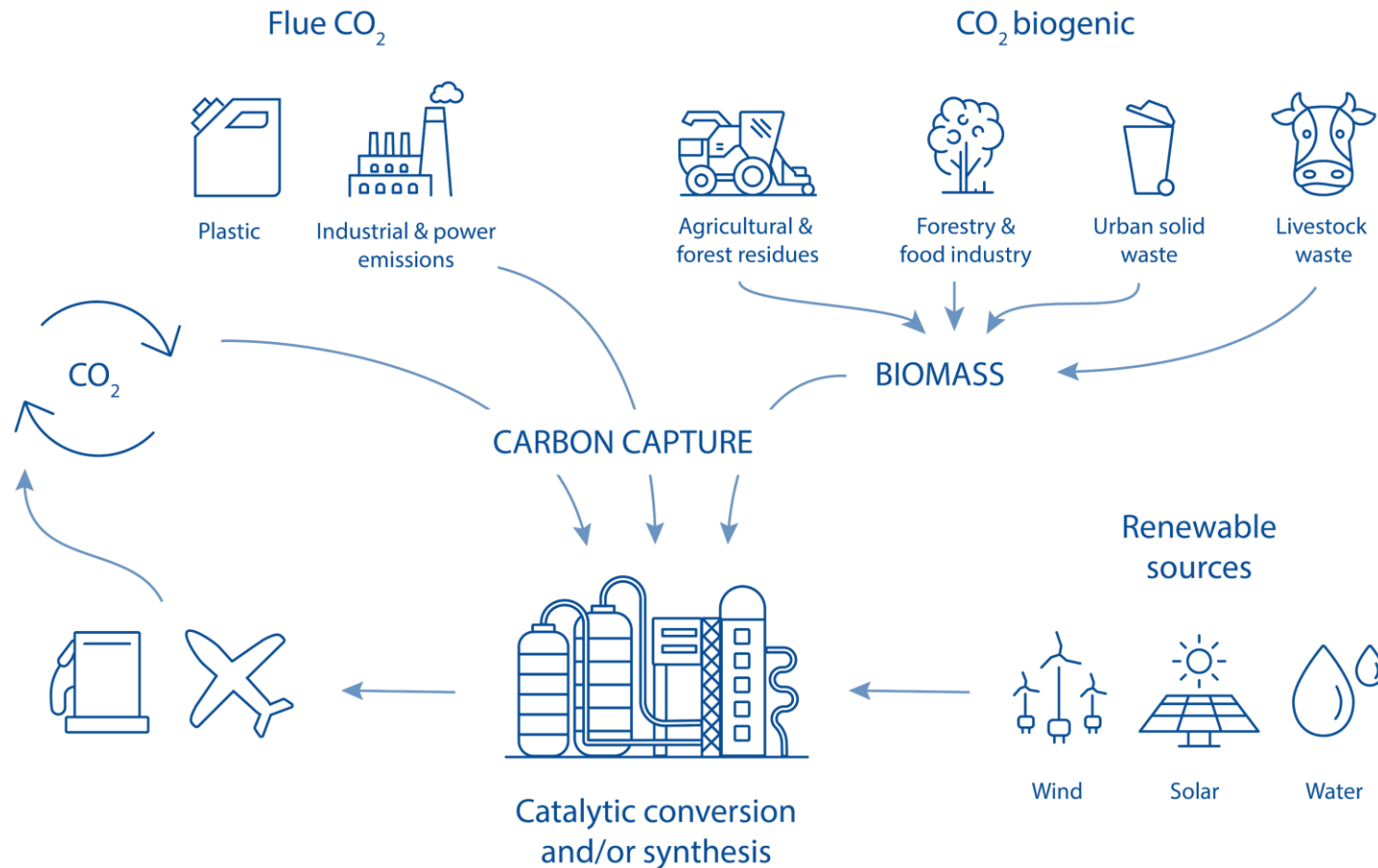
SAF production approach based on ASTM D7566



Unfortunately, **conventional technologies** often suffer from **low selectivity control** and **conversion** while **lacking energy efficiency**.

Therefore, **new technology** solutions are required, in which the **rational design of catalytic materials** is a must.

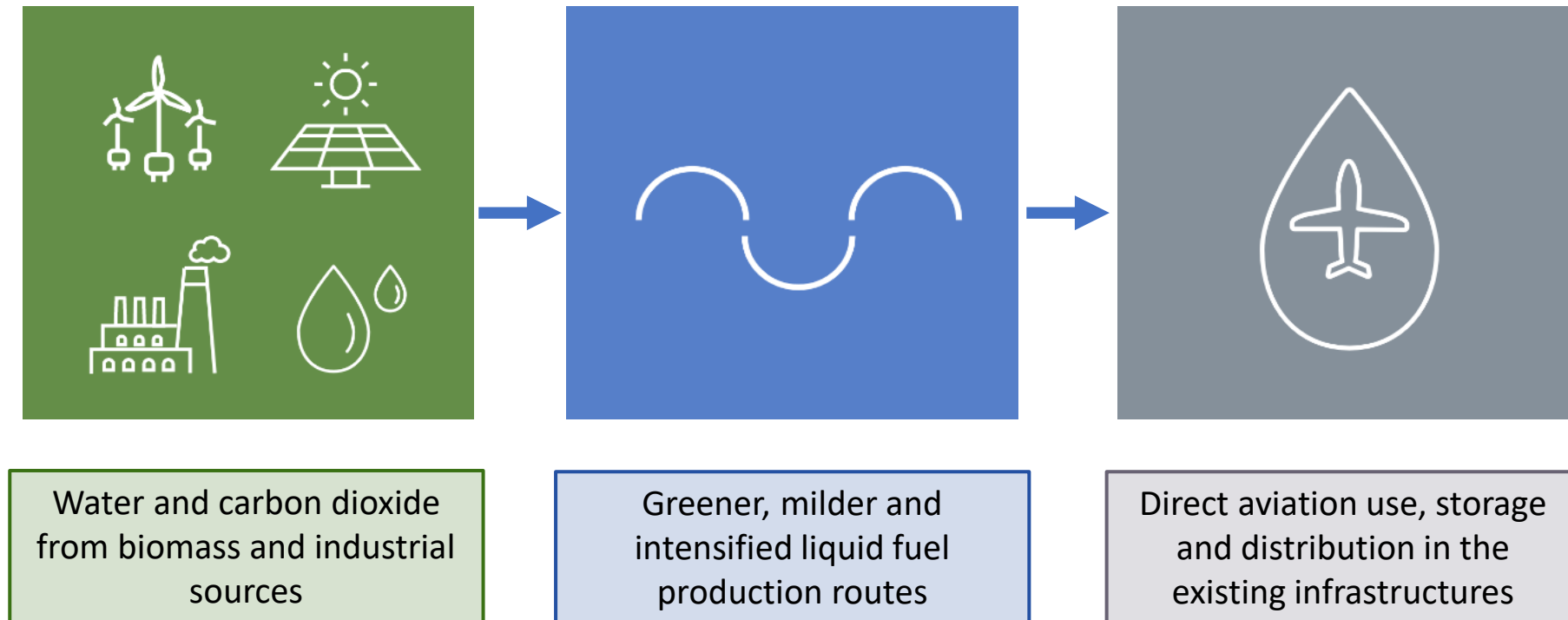
4AirCRAFT Approach



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4AirCRAFT Approach



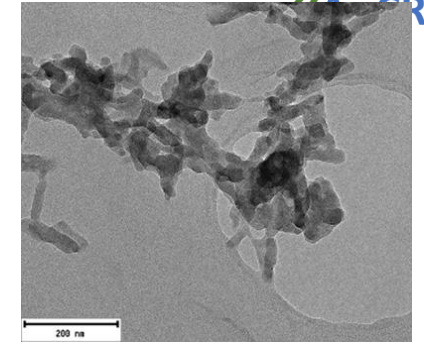
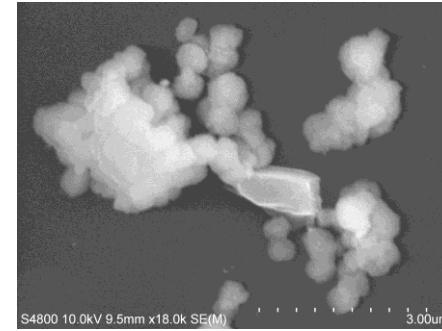
Hybrid cascade reactor technology - CO₂ conversion to long-chain hydrocarbons at mild conditions
Proof of the concept → TRL3

4AirCRAFT – Research Activities



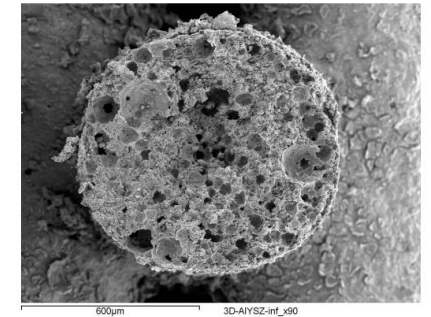
Materials

- Electrocatalyst
- Chemocatalysts
- Bio- and Biomimetic catalysts



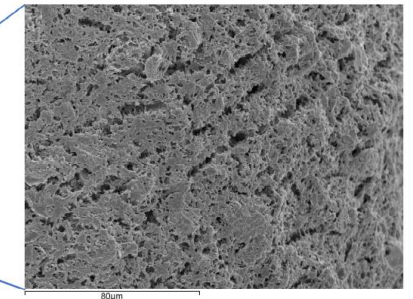
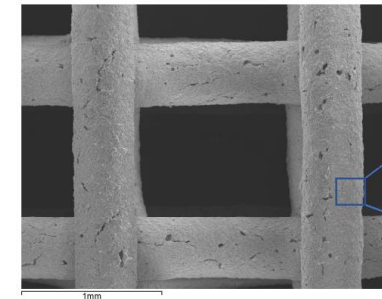
Components

- Membranes and Electrodes
- Advanced Catalysts Carriers (MOFs and nano→meso→macro structured and functionalized scaffolds)



Reactor

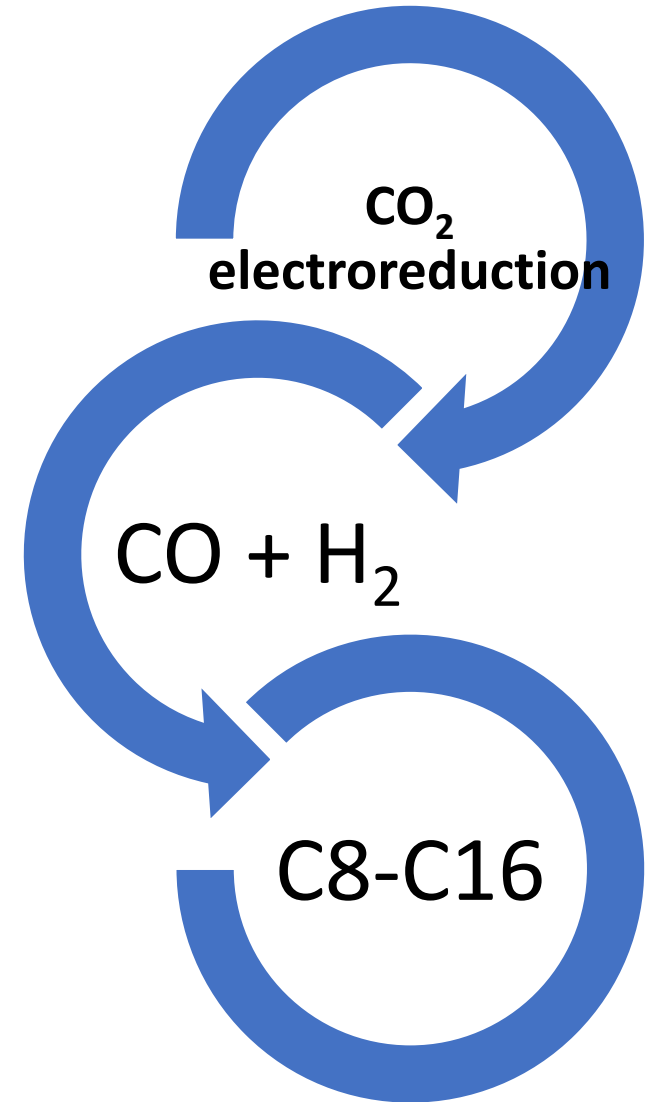
- Reactor design-Process Intensification
- Structural and mechanistic investigations
- Proof of the concept and Life Cycle Assessment



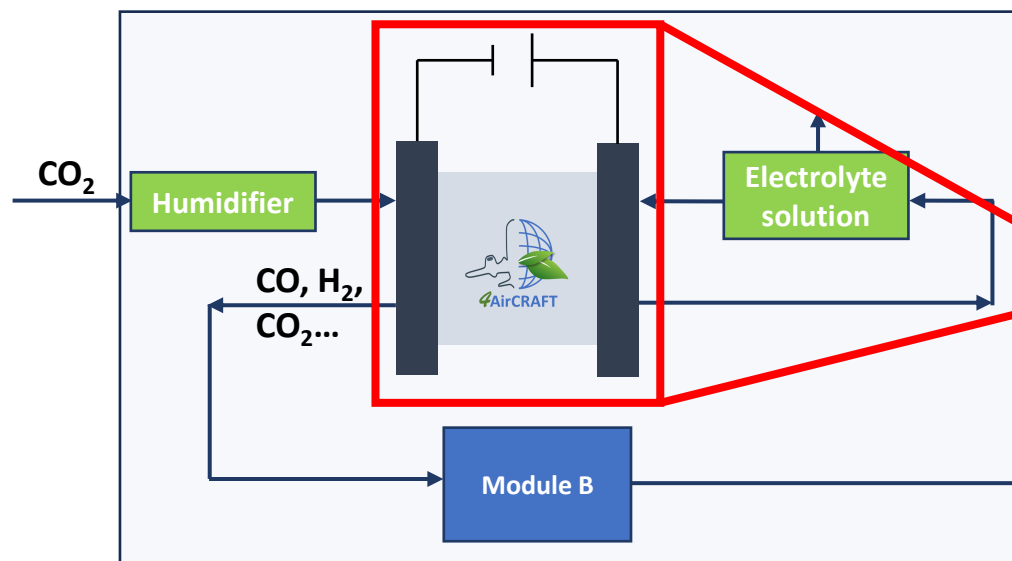


Air Carbon Recycling for Aviation Fuel Technology

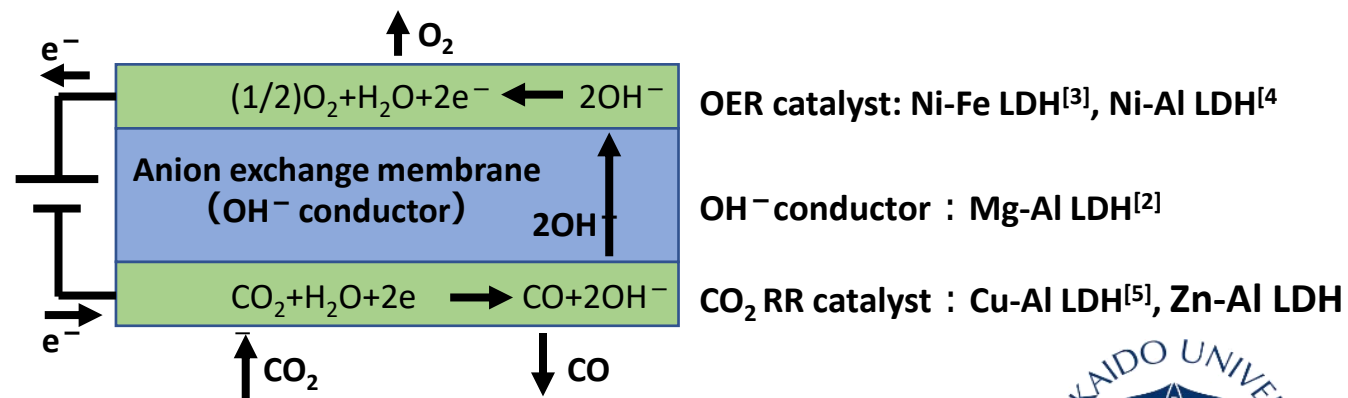
Preliminary results strategy “syngas mediated”



4AirCRAFT cascade reactor – key step CO production



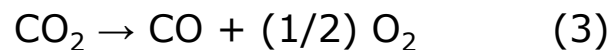
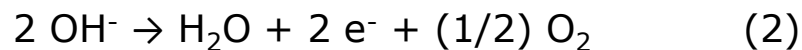
Electrochemical cell using an anion exchange membrane



LDH (Layered Double Hydroxide)

- Variety of metal combination^[1]
- Large surface area^[1]
- High OH⁻ conductivity^[2]
- Stability in alkaline solution^[2]

[1] C. I. Ezech et al., *Ultrason. Sonochem.* **2018**, 40, 341.
 [2] K. Tadanaga et al., *Adv. Mater.* **2010**, 22, 4401.
 [3] D. Zhou et al., *Chem. Soc. Rev.* **2021**, 50, 8790.
 [4] M. Li et al., *J. Mater. Sci.* **2019**, 54, 9034.
 [5] K. Iwase et al., *ChemSusChem* **2022**, 15, e202102340.



K. Tadanaga et al. *J Asian Ceram Soc.* **2023**, 11.3, 406



Prof. K. Tadanaga

Electrocatalyst performance - Cell configuration for CO₂ reduction reaction (CO₂RR)

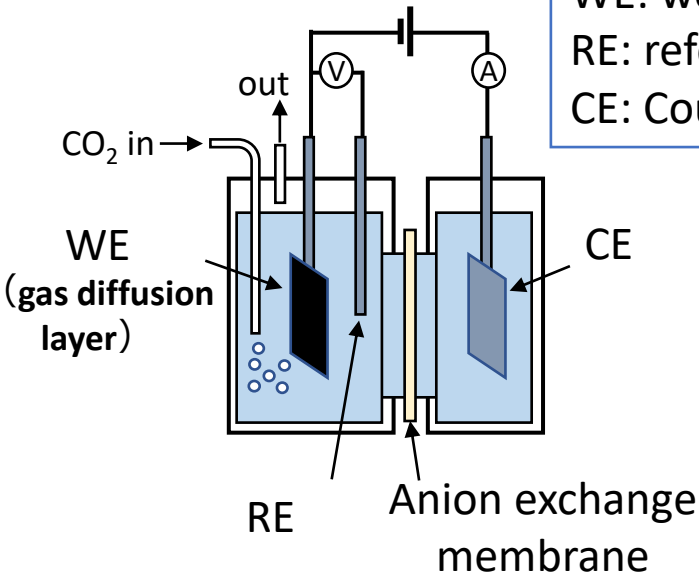


2-compartment cell

3-compartment cell

Zero gap cell

WE: working electrode
RE: reference electrode
CE: Counter electrode



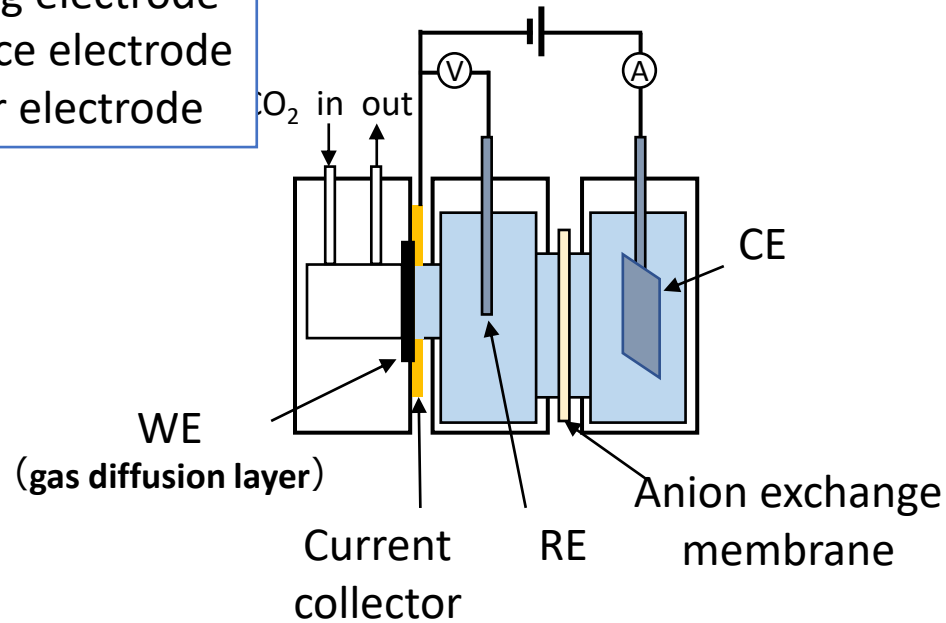
Dissolved CO₂RR

✓ Factors to limit the CO₂ RR reaction

Saturation concentration of CO₂ (33 mM)

Diffusion rate of CO₂

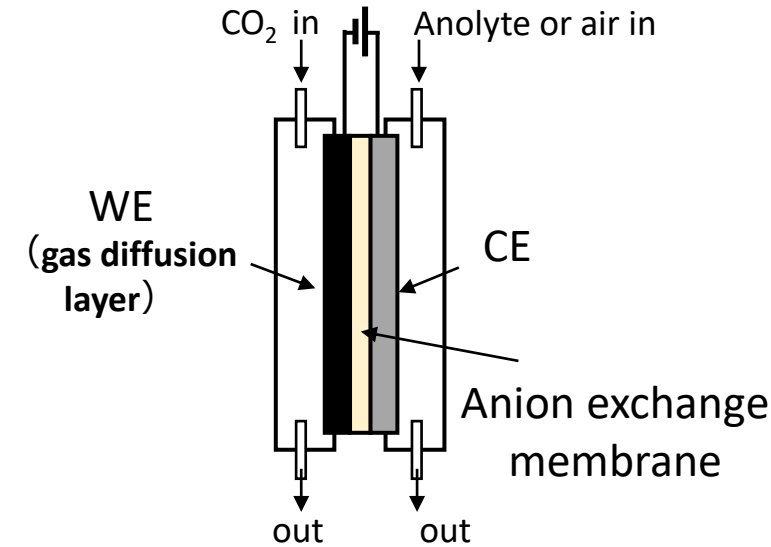
Limit of electrolyte because of dissolution of CO₂



Gaseous- CO₂RR

✓ Increase in reaction efficiency using gas

✓ Configuration is rather easy



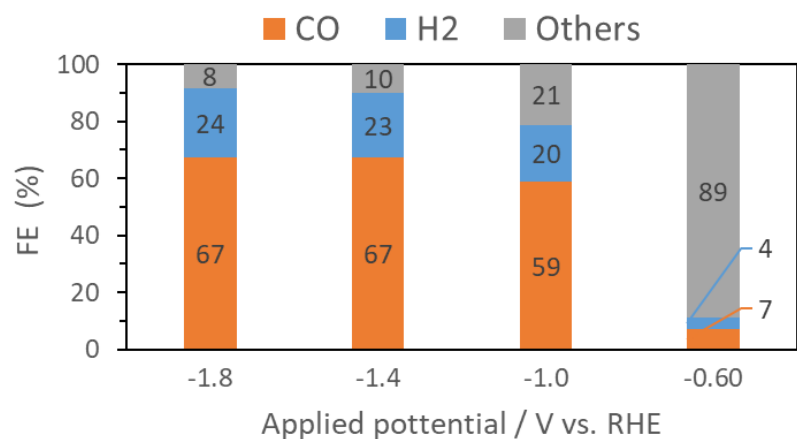
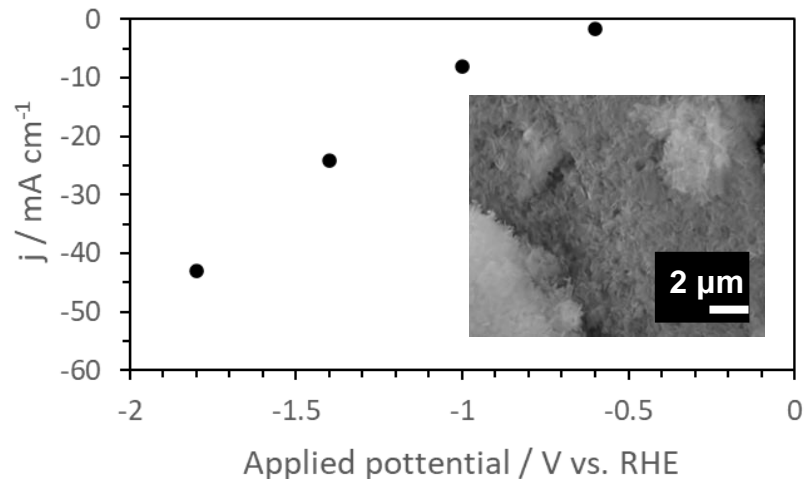
Practical gaseous CO₂RR

✓ Reduction of cell resistance using MEA (membrane electrode assembly)

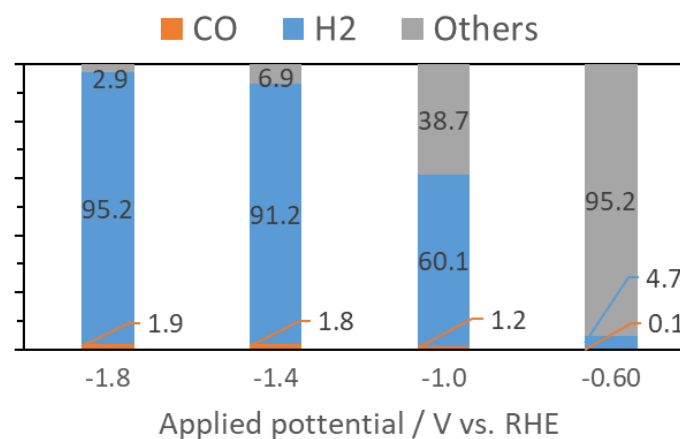
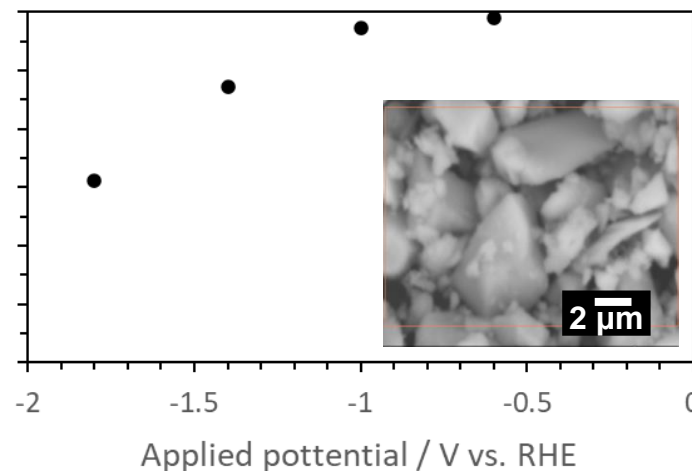
Gaseous CO₂RR activity

2-compartment cell test

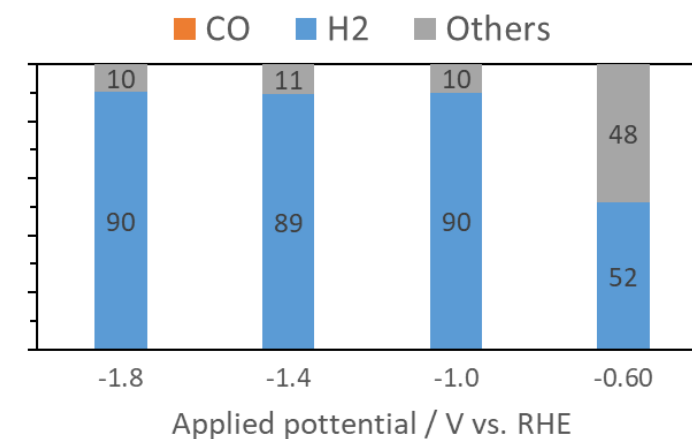
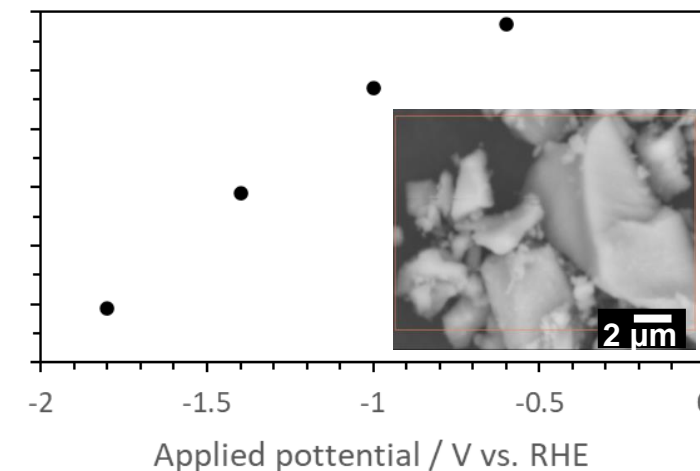
Zn-Al LDH



Ni-Al LDH

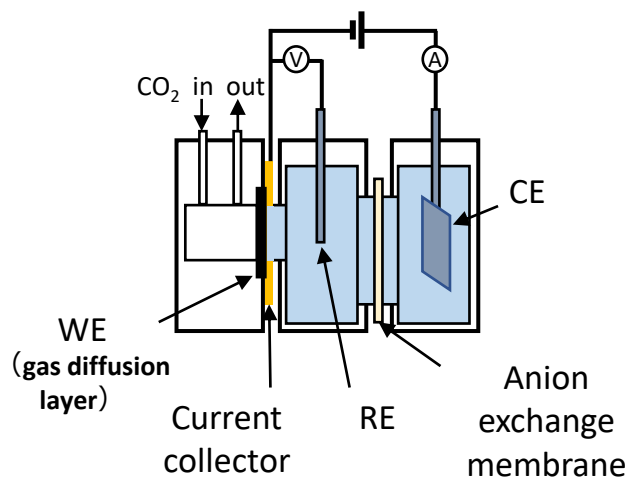


Ni-Fe LDH



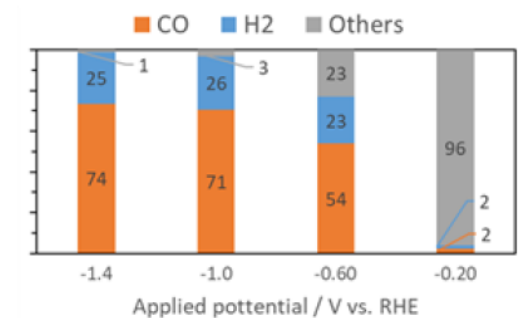
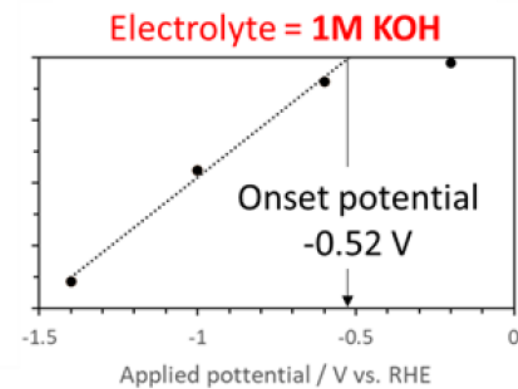
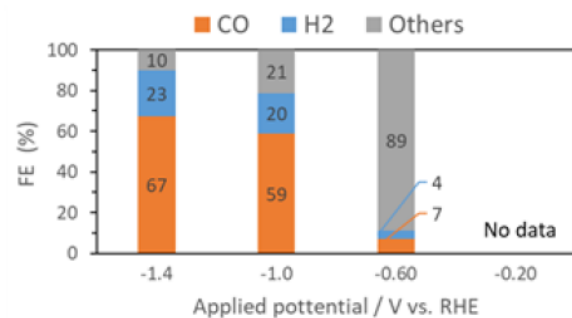
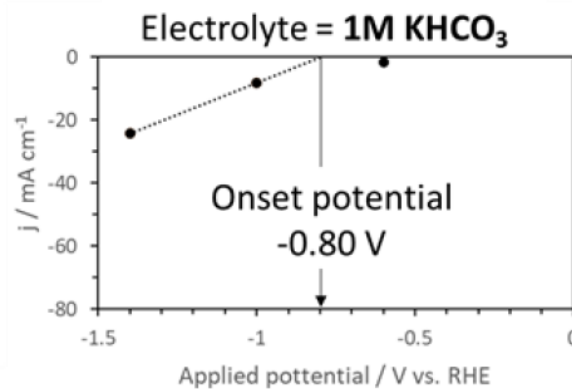
Gaseous CO₂RR activity

3-compartment cell test



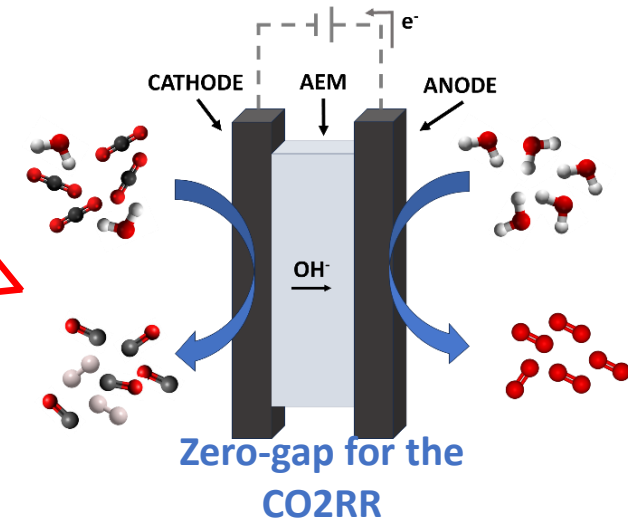
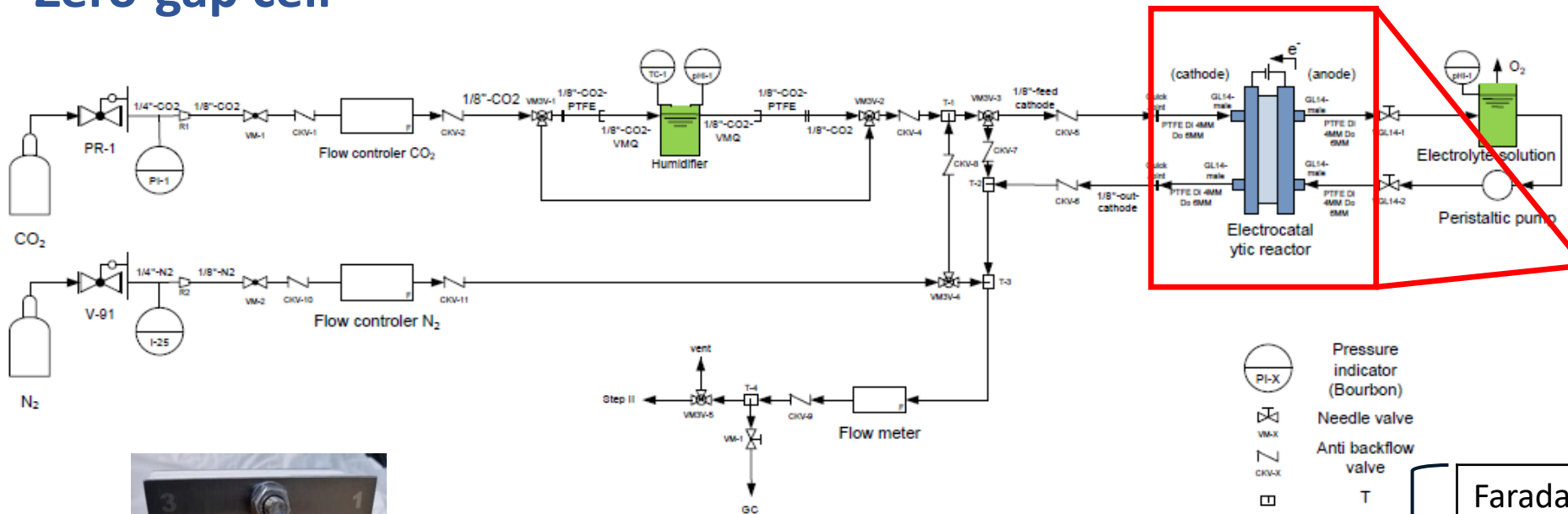
Gaseous- CO₂RR

- ✓ Increase in reaction efficiency using gas
- ✓ Configuration is rather easy

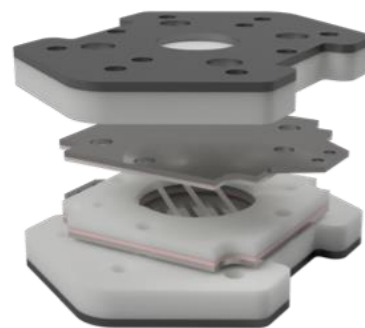


CO2RR Test Bench

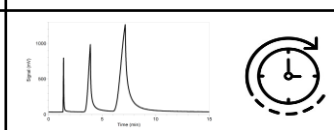


Zero-gap cell



Commissioning module – single cell



Initial module design – single cell

Faradaic efficiency	FE
Lifetime studies	
LCA	
TEA	

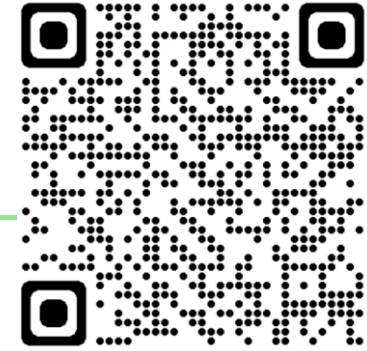
Conclusions



- ✓ Zn-based catalyst with CO₂ affinity achieves up to 77% of selectivity for CO evolution and 94% for CO + H₂.
- ✓ Design and construction of a test bench to carry out tests and monitoring carbon dioxide flues.
- ✓ Customizable electrochemical cell.
- ✓ Use of non-precious metal as catalysts.
- ✓ Development of a stage with great potential for the synthesis of industrial chemical processes.



Financial support from the T13-23R (the Aragón Government).



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